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Enantioselective synthesis of (−)-Bifurcadiol: a natural antitumor marine product

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Abstract

The enantioselective synthesis of natural antitumor (−)-Bifurcadiol involving an alkylation key-step reaction is reported. © 1999 Published by Elsevier Science Ltd. All rights reserved.

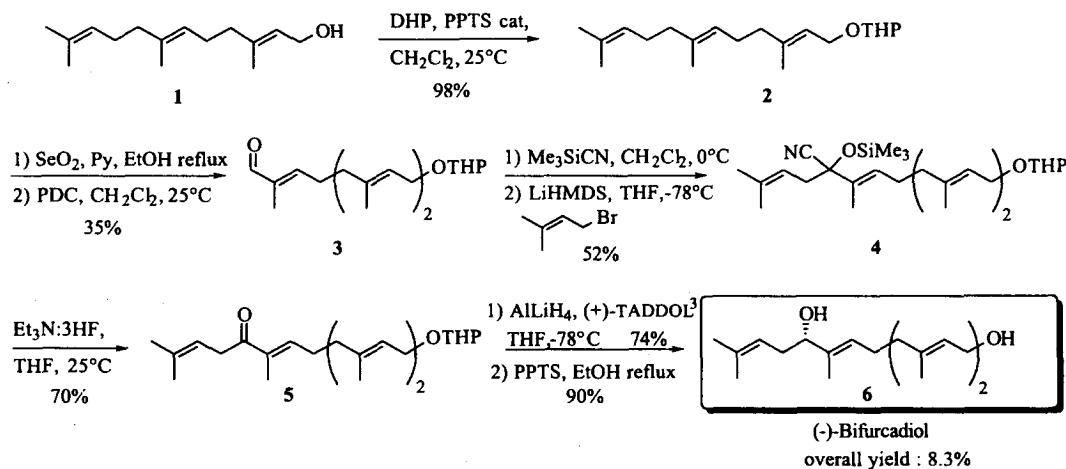
Keywords: marine natural product; (−)-Bifurcadiol; alkylation; asymmetric reduction.

In the course of our investigations on the brown alga *Bifurcaria bifurcata*, we have recently isolated a new class of di-functionalized oxygenated linear diterpenes.¹ Numerous biological activities such as antiulcer, antimutagenic or cytotoxic properties have been found among the linear diterpenes. *Bifurcaria bifurcata*, in particular, is a source of potent active compounds such as [12(S)-hydroxy-geranylgeraniol] ((−)-Bifurcadiol) exhibiting cytotoxicity against cultured human tumor cell lines (A549, SK-OV-3, SK-MEL-2, XF 498 and HCT 15).²

In this paper, we report the first enantioselective synthesis of (−)-Bifurcadiol **6** involving an alkylation key-step reaction between a cyanohydrin and an allyl bromide (Scheme 1).

trans trans Farnesol **1** was converted into the corresponding ether **2** by treatment with dihydropyran and subsequently oxidized using a mixture of pyridine, selenium dioxide and PDC to afford the expected aldehyde **3** in 34% yield.⁴ Compound **3** was added to a solution of trimethylsilyl cyanide in the presence of a catalytic amount of sodium cyanide and 18-crown-6 ether complex to lead, after addition to a solution of LiN(SiMe₃)₂ in anhydrous THF (−78°C) and condensation with 1-bromo-3-methylbut-2-ene, to the silyl cyanide derivative **4**⁵ (52% yield). Subsequent treatment with triethylamine trihydrofluoride salt afforded the expected ketone **5** in 70% yield.⁶

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Scheme 1.

The synthesis of natural (-)-Bifurcadiol was achieved involving an asymmetric reduction of ketone **5**. In a first attempt, an enantioselective borane reduction using Corey's oxazaborolidine catalyst (prepared from (*S*)-(-)-(diphenylhydroxymethyl)pyrrolidine and methylboronic acid) has been performed.⁷ Nevertheless, more than 95% of **5** has been recovered whatever the experimental conditions applied. The asymmetric reduction of **5** has been successfully realized using a stoichiometric AlLiH₄/(+)-TADDOL^{3,8} mixture affording, after removal of the protective group, (-)-Bifurcadiol **6** in 66% yield and 92% enantiomeric excess as determined by chiral HPLC analysis and measurement of optical deviation on a sample of the natural and synthetic corresponding diacetate.⁹ The same procedure has been applied using (-)-TADDOL¹⁰ to prepare the nonnatural (+)-Bifurcadiol in 73% yield and up to 99% enantiomeric excess.

Further studies involving this general method for the preparation of new linear di-functionalized oxygenated diterpenes with a potent biological interest, are in progress.

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- (a) ED₅₀ values ranging from 4.1–8.3 µg mL⁻¹. See: Zee, O. P.; Kim, D. K.; Choi, S. U.; Lee, C. O.; Lee, K. R. *Arch. Pharmacal. Res.* **1999**, *22*, 225–227.
- The (+)-TADDOL used is the (4*S*,5*S*)-2,2-dimethyl-α,α,α',α'-tetraphenyldioxolane-4,5-dimethanol.
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- The enantiomeric excesses were determined using a Daicel Merck LichroCART column 250-4 (*S,S*) Whelk-O1, refractometric detector, using a mixture of 2-propanol:hexane (5:95, v/v), flow rate 0.5 mL min⁻¹ and retention time of 10.2 min.
- The (-)-TADDOL used is the (4*R*,5*R*)-2,2-dimethyl-α,α,α',α'-tetraphenyldioxolane-4,5-dimethanol.